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SECTION 5.1 MODELS OF THE ATOM

(pages 127–132) This section

summarizes the development of  
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of electrons as they relate to the quantum mechanical model of the atom. The Development of Atomic Models (pages 127-128)

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## Development of Atomic Models

(pages 127-128) 1. Chapter 5

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## Atomic Orbitals Summary of

### Principal Energy Levels ... The Bohr

Model Period 5  $5s \rightarrow 4d \rightarrow 5p$  blocks

$5s^2 4d^{10} 5p^5$  Writing all the

sublevel blocks in order gives 1 s 2

2s 2 2p 6 3s 2 3p 6 4 s 2 3 d 10 4 p

6 5 s 2 4 d 10 5p 5 Chapter 5

## Electronic Structure and Periodic

### Trends According to Hund's rule,

when electrons occupy orbitals of

equal energy, one electron enters

each orbit until a. all the orbitals

contain one electron, with spins

parallel.

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