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space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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free Modern Chemistry answers. Shed the societal and cultural narratives holding you back and let free step-by-step Modern Chemistry textbook solutions reorient your old paradigms. Modern Chemistry Answer Key Chapter 10 - seapa.org CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ... 5 The Periodic Law Learn chapter 8 questions answers chemistry with free interactive

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overall ionic equation for the reaction. 32. Write the net ionic equation. Assessment Chapter Test B CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$

a. What is the value of the coefficient  $x$  in this equation? 40.07 g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol  $O_2$ :1 mol  $H_2O$  c.

What is the mole ratio ... mc06se cFMsr i-vi -

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Section 5 Review Answers...Section 5 Review p 207

Chapter Review p 209 Chapter 7 Chemical Formulas

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