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dilute aqueous solutions of the strong acid HI and the weak acid HCOOH. The solution of HI contains only H<sub>2</sub>O and I<sup>-</sup> and water molecules. The solution of HCOOH contains H<sub>3</sub>O<sup>+</sup> and HCOO<sup>-</sup> ions, and HCOOH and H<sub>2</sub>O molecules.

18. Relate the strength of a weak acid to the strength of its conjugate base.

Acids and Bases

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10. Possible answers include: white with blue lines, flat, thin, 1–2 g, flammable, solid.

11. Possible answers include: red/green/red & blue, cylindrical solid filled with liquid which has a high concentration of gas, 355 mL, pressure increases if shaken.

12. Possible answers include: color, rectangular. solid, made of paper and cardboard ... Teacher Guide & Answers - Glencoe

Section 18.1 • Introduction to Acids and Bases 635

Figure 18.2 The strong acid hydrochloric acid (HCl), also called muriatic acid, is used to clean bricks and concrete. The strong base sodium hydroxide (NaOH) can clear clogged drains. Acids turn blue litmus red. Bases turn red litmus blue.

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