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Words [2] (b) Marit arranged the values she found in
Table 12 into an energy cycle. $C_2H_4(g) + H_2(g)$
 $C_2H_6(g)$ 3 0 2 $2CO_2(g) + 3H_2O$ - 1 4 1 1 k J m o l - 1 -
2 8 6 k J m o l - 1 - 1 5 6 0 k J m o l - 1 (hydrogenation)
(l) Calculate the value for the enthalpy of
hydrogenation of ethene from the energy
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