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account both the nonzero volumes
of particles and the presence of
strong intermolecular attractive
forces. Solids and liquids have
particles that are fairly close to one
another, and are thus called
"condensed phases" to distinguish
them from gases 2.1: Condensed
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temperature is the key process
variable used to generate the
thermodynamic driving force for
transformation. For example, in
metal solidification the
nonisothermal conditions lead to
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bimolecular dynamics in solution

Study of the dynamics of bimolecular reactions in the liquid phase is a field that is in its infancy, but modern developments in laser instrumentation for generation of ultrafast (fs - ps) pump (UV) and transient IR probe pulses promise exciting new opportunities. The Reaction Dynamics Group The most familiar examples of condensed phases are solids and liquids, which arise from the electromagnetic forces between atoms. Condensed matter physicists seek to understand the behavior of these phases by using physical laws. In particular, they include the laws of quantum mechanics, electromagnetism and statistical mechanics. Condensed matter

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