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d. 3+ 43. a. 2+ b. 2+ c. 3+ d. 1+

44. a. barium ion b. iodide ion c. silver ion d. mercury(II) ion 45.

cyanide, CN^- and hydroxide, OH^-

46. a. hydroxide ion b. lead(IV) ion c. sulfate ion d. oxide ion 47. zero;

A compound is electrically neutral.

48. The symbols for the cation and

CHAPTER 9 Study Guide -

Quia Section 9.4 - Naming and

Writing Formulas for Acids and

Bases. An acid is a compound that produces H^+ ions when it dissolves

in water. The formula for an acid normally starts with and H. When

naming acids, you should first

determine whether or not the acid contains oxygen. Chapter 9 -

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- OBJECTIVES:

- Identify the charges on monatomic ions by using the periodic table, and name the ions.

3. Chemistry - Chp 9 - Chemical Names and Formulas - PowerPoint CHEMICAL NAMES AND FORMULAS SECTION 9.1 NAMING IONS (pages 253 - 258) This section explains the use of the periodic

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table to determine the charge of an ion. It also defines polyatomic ion and gives the names and formulas for the most common polyatomic ions. Monatomic Ions (pages 253-256) 1. What are monatomic ions? _____ 2. Unit 5: Chemical Names & Formulas Chapter 9 "Chemical Names and Formulas" Tools. Copy this to my account; E-mail to a friend; Find other activities; ... In samples of any chemical compound, the masses of the elements are always in the same proportions any atom or group of atoms that has a negative charge ... place cation name first followed by the anion name: naming polyatomic ... Quia - Chapter 9 "Chemical Names and Formulas" Get Free Chapter 9 Chemical Names And Formulas

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- Determine the charges of monatomic ions by using the periodic table and write the names of the ions.
- Define a polyatomic ion and write the names and formulas of the most common polyatomic ions.
- Identify the two common endings for the names of most polyatomic ions.

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7/9/04 3:29 PM Page 211 NAMING IONS 9 Download Free Chapter 9 Chemical Names And Formulas species of atom having the same number of protons in their atomic nuclei (that is, the same atomic number, or Z). For example, the atomic number of oxygen is 8, so the element oxygen describes all atoms which have 8 protons.. Chapter 9 Chemical Names And Formulas Chapter 9, Chemical Names and Formulas Item Classwork (check off when done) OK'd Item Homework (check off when done) OK'd Opening Activity Ionic and Molecular Compounds Complete the Pre -activity. Perform the activity - record data/observations in lab notebook. Complete the Post -activity assignment. Section s 9 .1 &

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