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then its concentration can be reported as 10 percent. Chapter 7 Acids, Bases, and Solutions Chapter 7 acids and bases 1. Chapter 7 Acid-Base Equilibria 2. Acid-Base Theories • Arrhenius theory (Nobel Prize) An acid is any substance that ionizes (partially or completely) in water to give hydrogen ions,  $H^+$  (which associate with the solvent to give hydronium ions,  $H_3O^+$ )  $HA + H_2O \rightleftharpoons H_3O^+ + A^-$  • A base: any substance that ionizes in water to give hydroxyl ions  $OH^-$ . Chapter 7 acids and bases -

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blue litmus paper red while bases

turn red litmus paper blue. Question

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