

Chapter 12 Stoichiometry Test B

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1. The first step in most stoichiometry problems is to _____. a. add the coefficients of the reagents c. convert given quantities to volumes b. convert given quantities to moles d. convert given quantities to masses _____

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mail.trempealeau.net Stoichiometry, pp. 74-83 TEST A
1. a 2. b 3. a 4. 5. d 6. c 7. a 8. d 9. b 10. d 11. a 12. c
13. b 14. a 15. b 16. d 17. a 18. c 19. c 20. b 21. d 22.
a 23. d 24. a 25. c TEST B 1. c 2. b 3. d 4. a 5. a 6. 7. a
8. b 9. given: mass of NH_3 500 g unknown: mass of N_2
10. given: mass of H_2O 500 g unknown: moles of H_2
11. given: amount of CO_2 20 ... Assessment Chapter
Test B - Ed W. Clark High School the maximum amount
of products that could be formed from given amounts
of reactants a. percent yield b. limiting reagent c.
theoretical yield d. stoichiometry e. actual yield f.
excess reagent B. Multiple Choice [Solved] the
maximum amount of products that could be ... Chapter
12 Test A A. Matching 1. b 3. e 5. a 2. c 4. f 6. d B.
Multiple Choice 7. b 11. b 15. c 8. c 12. c 16. a 9. b 13.
b 17. c 10. d 14. a 18. c C. Problems 19. There is no
limiting reagent, because the mole ratio of the
reactants is 1 mol N_2 to 3 mol H_2 . 20. 21. 15.0 g CO_2
1 4 m 4g ol C C O O 2 2 2 8 m m o o | | C C 4H O 2 10 1
5 m 8g ol ... Chapter 11 Small-Scale Lab Stoichiometry
Chapter 12 Test B Answers Stoichiometry Chapter 12
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Stoichiometry Chapter 12 Test B Answers and
collections to check out. 00 grams of hydrogen gas.
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chemistry. Khan Academy Introduction to ...
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